

Assessment of Water Quality of Udebal Lake, Nearby Santram Mandir, Nadiad, Dist. Kheda, Gujarat, India

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ABSTRACT

Assessment of Water Quality of Udebal Lake Near by Santram Mandir. It's Physico Chemical analysis such as, pH, Total Alkalinity, Total Hardness, T.D.S, Calcium, Magnesium, Chloride, Nitrate, Sulphate, D.O during January 2020 to January 2022, The present study calculate the Water Quality Index of lake and assess the impact of industries, agriculture and human activities. W.Q.I has been calculated of Udebal Lake, Nearby Santram Mandir, District: Kheda, Gujarat (India). Four different locations were selected in Udebal Lake, Nearby Santram Mandir. Total Ten Physico– Chemical parameters were monitored for the calculation of W.Q.I in Rainy, Winter and Summer seasons. The parameters namely pH, Total Alkalinity, Calcium, Magnesium, Chloride, Sulphate, were exceeding the permissible limits as prescribed by Indian Standards. However, The W.Q.I values in the present investigation were reported to be more than 60(68.89,113.958,138.74), Indicating that the Water Quality is poor in Rainy and Winter season and more poor in Summer season and used for drinking purpose after purification .The Pollution load is relatively lighter during Summer season which compare to the Winter and Rainy Season. Results obtained are compared in term of their highest value and lowest values among three seasons in term of 10 parameters.

KEYWORDS: Udebal Lake, Assessment, W.Q.I, Physico– Chemical, Season, Permissible Limits

1. INTRODUCTION

Water is extremely elementary to light one cannot imagine a form of life that might exist water without water on the surface of the earth Water in the form of ocean, sea, glaciers freshwater bodies, river Where Lake etc. occupies about 71.00 % of the area while, the landmass occupies about 29.00 % of the area [1,2]. Water is most essential for existence of life on earth and is a major component for all forms of life from micro-organism to men Various Physico-Chemical parameters have a significant role in determining the portability of Water. The use of fertilizers, pesticides and manure are main source of Water pollution in this area Water is one of the most important factor for every living organism on this planet Water is generally used for drinking fisheries and other domestic purpose in the in this area.

Therefore we carried out studies of Physico – Chemical parameters of surface water in Gujarat state

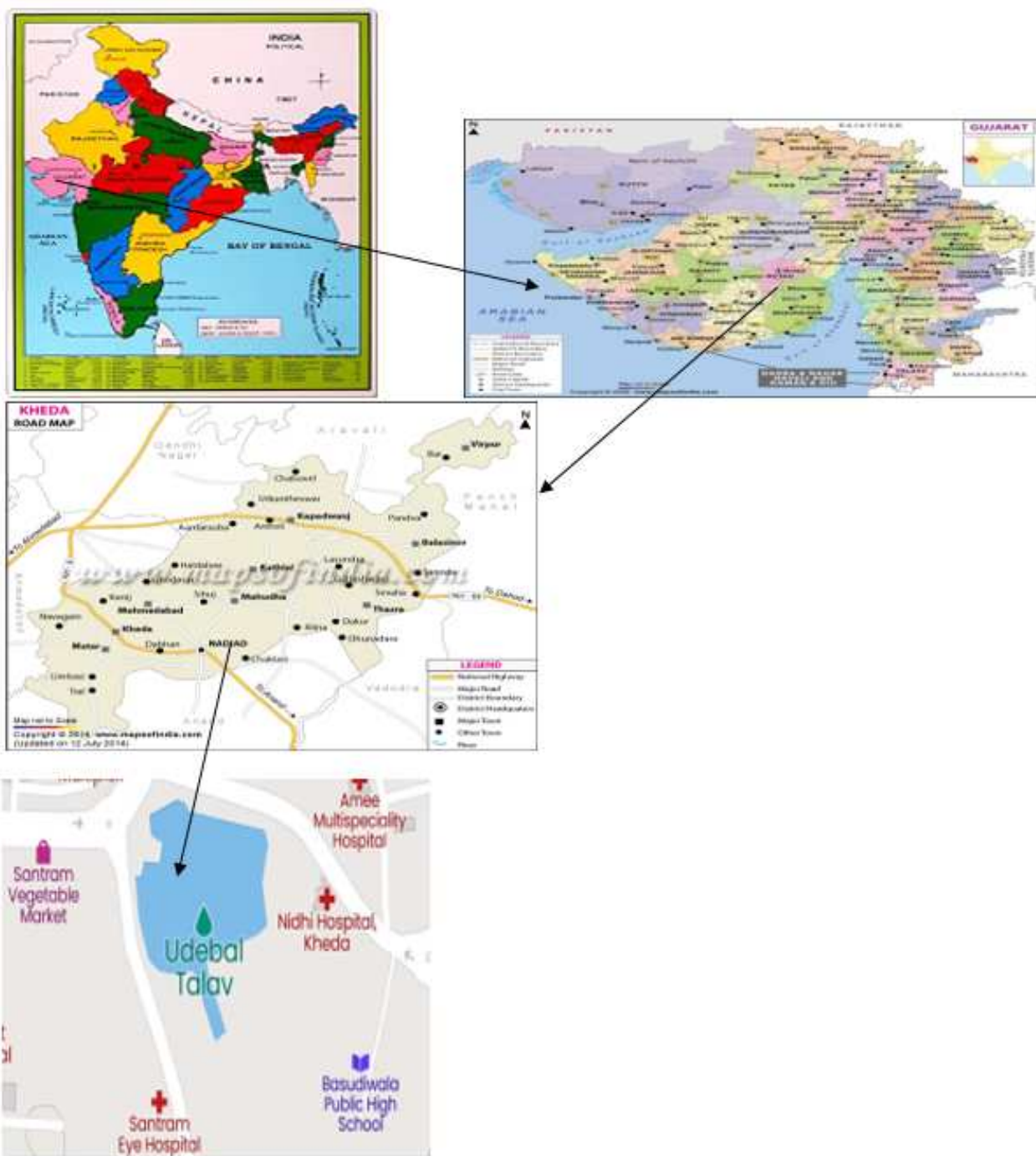
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Index based on some very important parameters can provide a single indicator of Water quality.



2. Methodology:

2.1. Study Area:

Udebal Lake is located Near by Santram Mandir in Nadiad. Udebal Lake is located between 22.6902° latitude and 72.8949° longitude. It is situated in Nadiad and comes under Kheda district. Udebal Lake Near by Santram Mandir is an old lake.

2.2. Samples Collections:

Samples of Water were collected from four sites of the lake once every month (January 2020 to January 2022). One liter P.E.T. bottles were used for collection of Water samples from a depth of 30 cm. During morning hours between 8.00 – 10.00 AM. For

Dissolved Oxygen measurement, a 300 ml capacity BOD bottle was used for collection of water sample and the Oxygen was fixed at the sampling site before being carried to the laboratory. The parameter pH was monitored at the sampling site and other parameters like Total Alkalinity, Total Hardness, Total Dissolved Solids, Calcium, Magnesium, Chloride, Nitrate and Sulphate were analyzed in the laboratory as per the standard procedures. APHA [2005]- [7,8].

The concept of Water Quality Index was first proposed by Horton (1965). For the calculation of Water Quality Index. 10 important Physico – Chemical parameters were chosen.

W.Q.I has been calculated by using the standards of drinking water quality recommended by the World Health Organization (*WHO*) – 1992 [9]. Bureau of Indian Standards (*BIS*) - 1993 and Indian Council for Medical Research (*ICMR*) - 1975. The weighted arithmetic index method (Brown et. al.) has been used for the calculation of *W.Q.I* of the lake. Further quality rating or sub index (*qn*) was calculated using the following expression.

$$qn = 100 \frac{[Vn - V_{10}]}{[Sn - V_{10}]}$$

Where,

qn = Quality rating for the *n*th water quality parameter.

Vn = Estimated value of the *n*th parameter at a given sampling station.

Sn = Standard permissible value of the *n*th parameter.

*V*₁₀ = Ideal value of *n*th parameter in a pure water.

Ideal value in most cases *V*₁₀ = 0 except in certain parameters like *P^H* and Dissolved Oxygen. Calculation of quality rating for *P^H* and *DO* (*V*₁₀ ≠ 0) is 7.0 and 14.6 mg/L respectively.

Unit weight was calculated by a value inversely proportional to the recommended standard values *Sn* of the corresponding parameters.

$$Wn = K / Sn.$$

Where, *Wn* = Unit weight for the *n*th parameter.

Sn = Standard value for *n*th parameter.

K = Constant for proportionality.

The overall Water Quality Index (*W.Q.I*) was calculated by aggregating the quality rating with the unit weight linearly.

$$\therefore W.Q.I. = \frac{\sum qnWn}{\sum Wn}$$

Table 1 Water Quality Index (*W.Q.I.*) and status of water quality (Chatterji and Raziuddin 2002)

Water Quality Index	Water Quality Status
0 – 25	Excellent Water Quality
26 – 50	Good Water Quality
51 – 75	Poor Water Quality
76 – 100	Very Poor Water Quality
> 100	Unfit for drinking

Table 2 Method used for Physico-Chemical analysis of water (All values except *pH* is in mg/L.)

Parameter (Unit)	Method
<i>pH</i>	<i>pH Probe</i>
Total Alkalinity	Volumetric
Total Hardness	Volumetric
<i>T.D.S.</i>	Gravimetric
Calcium	Volumetric
Magnesium	Volumetric
Chloride	Volumetric
Nitrate	Colorimetric
Sulphate	Volumetric
<i>D.O.</i>	Volumetric

Table – 3 Drinking water standards recommending agencies and unit weight. (All values except *pH* is in mg/L.)

Parameter	Standards	Recommended Agency	Unit Weight
<i>pH</i>	6.5 - 8.5	<i>ICMR / BIS</i>	0.2190
Total Alkalinity	120	<i>ICMR</i>	0.0155
Total Hardness	300	<i>ICMR / BIS</i>	0.0062
<i>T.D.S.</i>	500	<i>ICMR / BIS</i>	0.0037
Calcium	75	<i>ICMR / BIS</i>	0.025
Magnesium	30	<i>ICMR / BIS</i>	0.062
Chloride	250	<i>ICMR</i>	0.0074
Nitrate	45	<i>ICMR / BIS</i>	0.0413
Sulphate	150	<i>ICMR / BIS</i>	0.0124
<i>D.O.</i>	5	<i>ICMR / BIS</i>	0.723

Table 4 Seasonal variations of the Physico – Chemical parameters of the Udebal Lake, Nearby Santram Mandir. (All values except *P^H* is in mg/L.)

Parameter	Seasons		
	Rainy Season	Winter Season	Summer Season
<i>pH</i>	7.49	8.7	8.9
Total Alkalinity	178	211.2	521
Total Hardness	172	233.6	485
<i>T.D.S.</i>	446	800	977
Calcium	14	72	149

Magnesium	28	80	127
Chloride	72	120	300
Nitrate	24	24	52
Sulphate	8	12	23
D.O.	6.9	4	3
Water Quality Index	68.89	113.958	138.34

Table 5 Calculation of Water Quality Index in Rainy Season

Parameter	Observed Values (Vn)	Standard Values (Sn)	Unit Weight (Wn)	Quality Rating (Qn)	WnQn
pH	7.49	6.5 - 8.5	0.2190	32.66	7.15254
Total Alkalinity	178	120	0.0155	148.333	2.2991
Total Hardness	172	300	0.0062	57.333	0.3554
T.D.S.	446	500	0.0037	89.2	0.33
Calcium	14	75	0.025	18.666	0.46665
Magnesium	28	30	0.062	93.33	5.7864
Chloride	72	250	0.0074	288	0.21312
Nitrate	24	45	0.0413	53.33	2.2025
Sulphate	8	150	0.0124	5.333	0.06612
D.O.	6.9	5	0.723	80.20	57.98
			$\sum Wn = 1.1155$	$\sum Qn = 607.185$	$\sum WnQn = 76.85183$
Water Quality Index = $\frac{\sum WnQn}{\sum Wn} = 76.85183 / 1.1155 = 68.89$					

Table – 6 Calculation of Water Quality Index in Winter Season

Parameter	Observed Values (Vn)	Standard Values (Sn)	Unit Weight (Wn)	Quality Rating (Qn)	WnQn
pH	8.7	6.5 - 8.5	0.2190	100	21.9
Total Alkalinity	211.2	120	0.0155	176	2.728
Total Hardness	233.6	300	0.0062	77.866	0.4827
T.D.S.	800	500	0.0037	160	0.592
Calcium	72	75	0.025	96	2.4
Magnesium	80	30	0.062	266.66	16.53
Chloride	120	250	0.0074	48	0.3552
Nitrate	24	45	0.0413	53.33	2.2025
Sulphate	12	150	0.0124	8	0.0992
D.O.	4	5	0.723	110.4166	79.8312
			$\sum Wn = 1.1155$	$\sum Qn = 1096.27$	$\sum WnQn = 127.1208$
Water Quality Index = $\frac{\sum WnQn}{\sum Wn} = 127.1208 / 1.1155 = 113.9585$					

Table – 7 Calculation of Water Quality Index in Summer Season

Parameter	Observed Values (Vn)	Standard Values (Sn)	Unit Weight (Wn)	Quality Rating (Qn)	Wn/ Qn
pH	8.9	6.5 - 8.5	0.2190	100	21.9
Total Alkalinity	521	120	0.0155	434.16	6.7294
Total Hardness	485	300	0.0062	161.66	1.0022
T.D.S.	977	500	0.0037	195.4	0.7229
Calcium	149	75	0.025	198.66	4.966
Magnesium	127	30	0.062	423.33	26.24
Chloride	300	250	0.0074	120	0.888
Nitrate	52	45	0.0413	115.55	4.7722

Sulphate	23	150	0.0124	15.33	0.19
D.O.	3	5	0.723	120.83	87.36
			$\sum W_n = 1.1155$	$\sum Q_n = 1884.92$	$\sum W_n Q_n = 154.7707$
Water Quality Index = $\frac{\sum W_n Q_n}{\sum W_n} = 154.7707 / 1.1155 = 138.74$					

3. Summary and Discussion:

The Physico-Chemical parameters of Water quality were analyzed using standard methods given in APHA, et al (2005). The values of various Physico – Chemical parameters for calculation of Water Quality Index are presented in Table: 4. While season wise Water Quality Index calculations are depicted in the Table 5, 6 and 7. The Water Quality Index of Rainy season, Winter season and Summer season are 68.59, 113.958 and 138.74 respectively. Which indicate that the Water quality is very poor to unfit for drinking used for human consumption after purification system (Chatterji and Raziuddin 2002)- [9]. It is also observed that the pollution load is relatively higher during Summer season when compared to the Winter and Rainy season. The above water quality is also supported by the following Physico – Chemical parameters variations observed during the different seasons of the study.

3.1. pH

pH is a measurement of the acidic or basic Quality of water[10,11]. The average pH values of the lake water was 7.49 mg/L during rainy season, 8.7 mg/L during winter season and 8.9 mg/L during summer season. The P^H of water was relatively high in the Summer season and low in Rainy and Winter season. However, when the average values for three seasons are taken into account that the water body was slightly alkaline[12]. Swaranalatha and Narasingrao [1993]- [13]. The P^H values of water samples of present study ranged from 7.49 to 8.9 of Udebal Lake Near by Santram Mandir. These values are within the prescribed limit of standards (WHO – 2002).

3.2. Total Alkalinity

Alkalinity value greater than 178 mg/L is desirable for domestic use APHA (2005). The observed average value of total alkalinity was 178 mg/L during Rainy season, 211.2mg/L during Winter season and 521mg/L during summer season. Total alkalinity values in our observations indicated that the water was hard. Higher values of alkalinity registered during Summer season, lower during Winter and Rainy season. D. K. Bhoi et. al. (2011) also reported similar findings in the study of Pariyej lake.[14]

3.3. Total Hardness:

The observed average Total Hardness value was 172 mg/L during rainy season, 233.6 mg/L during winter season and 485 mg/L during Summer can be

attributed to low water level and high rate of evaporation of water and addition of Calcium and Magnesium salt. Mohanta and Patra (2000) stated that addition of sewage, detergents and large scale human use might be the cause of elevation of hardness[15,16]. Udebal Lake, Near by Santram Mandir water was moderately hard but the hardness values were up to permissible limits. Hardness below 300 mg/L is considered potable but beyond this limit produces gastrointestinal irritation (ICMR 1975).

3.4. Total Dissolved Solids:

The total Dissolved solids in water of Udebal Lake was 446 mg/L. during rainy season, 800 mg/L. during winter season and 977 mg/L. during Summer season. Gupta and Singh (2000) also reported high concentration of TDS in the Damodar river due to mixing of sewage and industrial water. Gray N. F. (2005) reported that the hardness in the water is due to the Dissolved minerals from sedimentary rocks, seepage and run-off. Detergents and soaps also aggravate the situation – Ahluwalia V. K. (2008)- [17,18]

3.5. Calcium:

The observed average value of Calcium was 14 mg/L during rainy season, 72 mg/L during winter season and 149 mg/L during Summer season. The quantities of Calcium in natural water depend upon the type of rocks. Small concentration of Calcium is reducing corrosion in water pipes- D. K. Bhoi et al., (2011)- [19].

3.6. Magnesium:

The observed average value of Magnesium was 28 mg/L during Rainy season, 80 mg/L during Winter season and 127 mg/L during Summer season. The values of Magnesium Winter, Summer season were not in permissible limits, Magnesium hardness particularly associated with the Sulphate ion has laxative effect on persons unaccustomed to it. (Khursid 1998)-. The recorded increase in Ca and Mg concentration during pre monsoon may be the effect of bacterial decomposition. Jaybhaye and Madlapure (2005) reported high values of Calcium and Magnesium in Parola dam, Dist. Hingoli, Maharashtra.

3.7. Chloride:

Chloride occurs in all types of natural waters. The high concentration of Chloride is considered to be an

indication of pollution due to high organic waste of animal origin (Singh, 1995). The observed Chloride value was 72 mg/L during rainy season, 120 mg/L during winter season and 300 mg/L in summer season. The higher values of Chloride recorded in summer season. Similar results were also reported by D.K.Bhoi et al., (2011) from Pariyej lake, Kheda district, Gujarat. The high values may be attributed to low water levels during Summer season- [20,21].

3.8. Nitrate:

The Nitrate ranged from 24 mg/L during rainy season, 24 mg/L during winter season and 52 mg/L during summer season during. In present study water sample of rainy and winter showed low concentration of Nitrate well below permissible levels in Rainy, Winter but higher in Summer as per the standards. According to Jhingram and Sugunan (1990) the water with the 0.2 to 0.5 ppm of Nitrate is of high productive reservoirs, upto 0.2 ppm Nitrate is medium productive reservoirs and in low productive reservoirs, the Nitrates are negligible. According to the above classification present reservoir belongs to high productive nature.

3.9. Sulphate:

Sulphate ion does not effect the taste of water if present in low concentration. The observed average value of Sulphate was 8 mg/L during rainy season, 12 mg/L during winter season and 23 mg/L during summer season. The Sulphate value in Udebal Lake water was found within the acceptable limit of 150 mg/L.

4. Dissolved Oxygen (D.O):

Dissolved Oxygen (D.O.) is important to the health of aquatic ecosystem because all aquatic organisms need Oxygen to survive. The average Dissolved Oxygen (D.O.) was 6.9 mg/L during rainy season, 4 mg/L during winter season and 3 mg/L during summer season. The maximum Dissolved Oxygen in the water of Udebal Lake was recorded in rainy season. Thereafter it started declining gradually and in summer reached the lowest concentration. The maximum DO in rainy season may be due to low atmospheric temperature and minimum D.O. was recorded in Summer season may be due to high metabolic rate of organism similar result was also reported by munawar [1970] – [16] and Bhoi et al. [2005] – [20] .

5. Conclusion:

The W.Q.I. values in the present investigation of Udebal Lake in rainy season, winter season and summer season are 68.89, 113.958 and 138.74 respectively. Which indicate that the water quality is poor to unfit for drinking.

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